

ASYMMETRIC DIELS-ALDER REACTION OF A CHIRAL ALLENIC ESTER :
ENANTIOSELECTIVE SYNTHESIS OF (-)- β -SANTALENE [1]

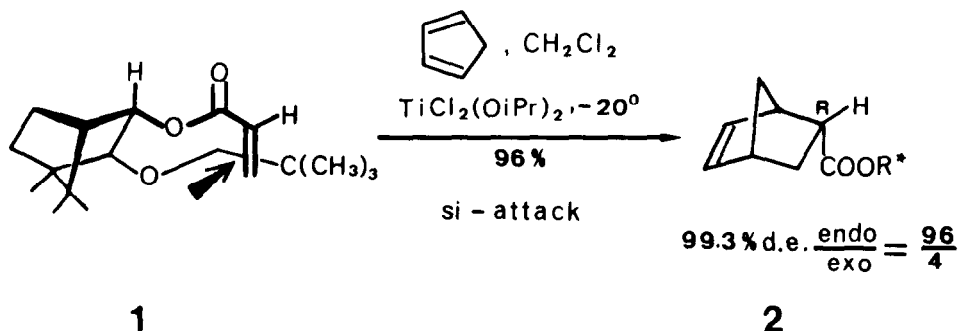
Wolfgang Oppolzer* and Christian Chapuis

Département de Chimie Organique, Université de Genève
CH-1211 Genève, Switzerland

Abstract. The Lewis-acid promoted Diels-Alder reaction of the allenic ester 10 provides the adduct 11 with excellent π -facial selection leading to an efficient enantioselective synthesis of (-)- β -santalene and to the recovery of the chiral control element 7.

Asymmetric Diels-Alder reactions using chiral dienophiles or dienes have received a great deal of attention over the last few years [2]. Thus, almost quantitative diastereoface differentiation has been achieved in the acrylate/cyclopentadiene addition 1 \rightarrow 2 as recently reported by this laboratory [3] (Scheme 1).

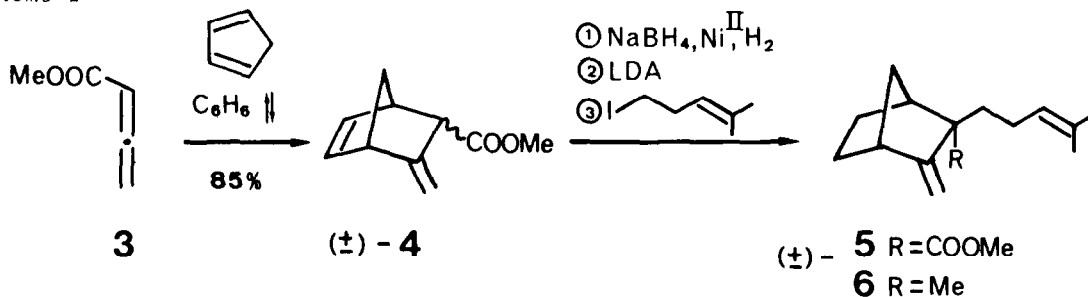
Scheme 1



The easy accessibility of the rationally designed, recoverable, camphor-derived control element, together with its predictable and efficient topological bias, offers attractive possibilities for the synthesis of chiral natural products.

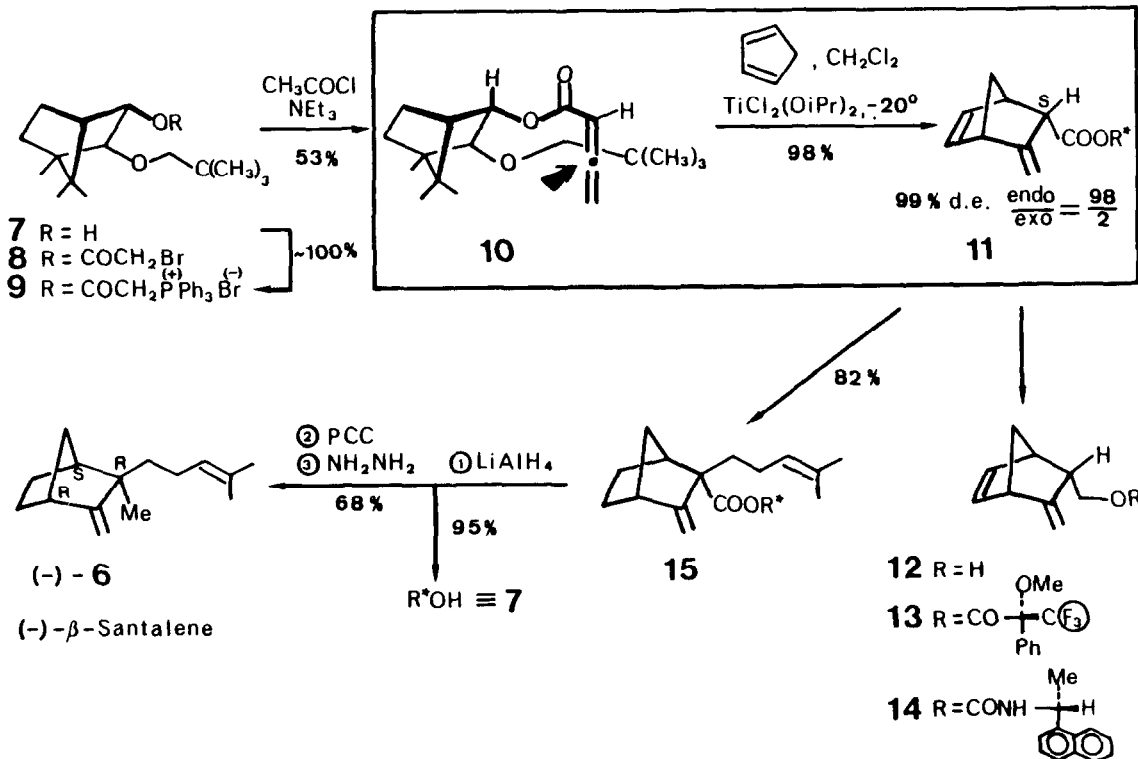
In this context, we present here a highly enantioselective total synthesis of the olfactively interesting sandalwood constituent (-)- β -santalene (**6**) [4]. Its racemate has been synthesized elegantly by *Bertrand et al.* [5] by means of a thermal allenic ester/cyclopentadiene addition **3** + **4** (Scheme 2) [6].

Scheme 2



Aiming at an efficient π -facial selection in the initial cycloaddition step, we have prepared the allenic ester **10** (Scheme 3).

Scheme 3



Acylation of the chiral alcohol 7 [3] with bromoacetyl bromide (2 eq., AgCN (1.4 eq.), benzene, 80°, 20 min) [7], followed by treatment of the resulting bromoester 8 [8] with triphenylphosphine (1.06 eq., benzene, 80°, 15 h) furnished after crystallization (CCl₄) the phosphonium salt 9 [8] (m.p. 183-185°, 100% yield from 7). Wittig-type reaction [9] of 9 with acetyl chloride (1 eq., NEt₃ (2 eq.), CH₂Cl₂, 0° → 40°) gave after distillation the allenic ester 10 [8] (b.p. 145°/0.15 Torr) in 53% yield.

Proceeding to the crucial cycloaddition step, treatment of 10 with cyclopentadiene (3 eq.) in the presence of TiCl₂(OiPr)₂ (1.5 eq., CH₂Cl₂, -20°, 6 h) [3] afforded the cycloadducts 11 [8] (oil) in 98% yield. To determine the diastereoisomeric purity of 11 it was reduced with LiAlH₄ (excess, Et₂O, 20°). The resulting *endo*-alcohol 12 [8] contained only 2% of its *exo*-epimer (GC) which was readily removed by preparative GC. Acylation of pure 12 with (R)-(+)- α -methoxy- α -trifluoromethylphenylacetic acid (1.5 eq., DCC (1.5 eq.), DMAP) and ¹⁹F-NMR analysis [10] of the resulting ester 13 in the presence of Pr(fod)₃ showed 12 to be enantiomerically pure. Alternatively, 12 was treated with (R)-(-)-1-(1-naphthyl)-ethyl isocyanate [11] in the presence of N-N-dimethyl-2-aminoethanol to give the carbamate 14. HPLC-analysis of 14 confirmed the 99% optical purity of 12. Accordingly, excellent chiral efficiency [12] has been achieved in the Lewis-acid promoted Diels-Alder addition 10 → 11. The absolute sense of induction follows from the conversion of 11 into (-)- β -santalene (6), carried out along the lines of the previously established synthesis of racemic 6 [5].

Selective hydrogenation of crude 11 (NaBH₄, Ni(OAc)₂, H₂), subsequent α -alkylation of the resulting ester [8] (1:LDA, 2:4-methyl-3-pentenyl iodide, THF/HMPA) and simple crystallization (EtOH) furnished the pure ester 15 [8, 13] (m.p. 119-120°) in 82% yield from crude 11. Reduction of 15 with LiAlH₄ (excess, Et₂O) refurnished the auxiliary element 7 and gave a crystalline alcohol [8] (m.p. 38-40°) which on successive oxidation (PCC, CH₂Cl₂) and Wolff-Kishner reduction (NH₂NH₂, KOH) afforded optically pure (-)- β -santalene (6). Synthetic (-)-6 was identified by comparison ([α] [13], IR, ¹H-NMR, ¹³C-NMR, MS) with (-)- β -santalene of natural origin. We believe that this work demonstrates convincingly that asymmetric C,C-bond-formation by π -facial selection of enoates has reached the stage of practical applicability [14].

Acknowledgements: Financial support of this work by the *Swiss National Science Foundation, Sandoz Ltd, Basel, and Givaudan SA, Vernier*, is gratefully acknowledged. We also thank Mr. J.P. Saulnier, Mr. A. Pinto and Mrs. D. Clément for NMR and MS measurements.

REFERENCES AND NOTES

- [1] To be presented at the 8th International Symposium "Synthesis in Organic Chemistry", Cambridge, England, July 1983.
- [2] a) *H.M. Walborsky, L. Barash, T.C. Davis, Tetrahedron* **19**, 2333 (1963); *J. Sauer, J. Kredel, Tetrahedron Lett.* **1966**, 6359; *R.F. Farmer, J. Hamer, J. Org. Chem.* **31**, 2418 (1966); *E.J. Corey, H.E. Ensley, J. Am. Chem. Soc.* **97**, 6908 (1975); *R.K. Boeckman Jr., P.C. Naegely, S.D. Arthur, J. Org. Chem.* **45**, 752 (1980); *D. Horton, T. Machinami, J. Chem. Soc. Chem. Commun.* **88** (1981); *G. Helmchen, R. Schmierer, Angew. Chem.* **93**, 208 (1981); *Angew. Chem. Int. Ed. Engl.* **20**, 205 (1981); *W. Oppolzer, M. Kurth, D. Reichlin, F. Moffatt, Tetrahedron Lett.* **2545** (1981); *W. Oppolzer, M. Kurth, D. Reichlin, C. Chapuis, M. Mohnhaupt, F. Moffatt, Helv. Chim. Acta* **64**, 2802 (1981); *W. Choy, L.A. Reed III, S. Masamune, J. Org. Chem.* **48**, 1137 (1983).
 b) *B.M. Trost, S.A. Godleski, J.P. Genêt, J. Am. Chem. Soc.* **100**, 3930 (1978); *B.M. Trost, D. O'Krongly, J.L. Belletire, Ibid.* **102**, 7595 (1980); *S. David, J. Eustache, A. Lubineau, J. Chem. Soc. Perkin I*, 1795 (1979); *W.G. Dauben, R.A. Bunce, Tetrahedron Lett.* **1982**, 4875; for an intramolecular asymmetric Diels-Alder reaction where the chirality-directing unit is attached to the chain which links the reaction partners see: *T. Mukaiyama, N. Iwasawa, Chem. Lett.* **29** (1981).
- [3] *W. Oppolzer, C. Chapuis, M.D. Guo, D. Reichlin, T. Godel, Tetrahedron Lett.* **1982**, 4781.
- [4] Constitution: *L. Ruzicka, G. Thomann, Helv. Chim. Acta* **18**, 355 (1935); configuration: *C.R. Eck, G.L. Hodgson, D.F. MacSweeney, R.W. Mills, T. Money, J. Chem. Soc., Perkin Trans I*, **1974**, 1938.
- [5] *M. Bertrand, H. Monti, K.C. Huong, Tetrahedron Lett.* **1979**, 15.
- [6] For other syntheses of santalenes and santalols see the reviews: *G. Buchbauer, Chemiker-Zeitung* **100**, 225 (1976); *E.-J. Brunke, E. Klein, in "Fragrance Chemistry", Ed. E.T. Theimer, Academic Press 1982, p. 397, and more recent work: K. Sato, O. Miyamoto, S. Inoue, K. Honda, Chem. Lett. **1981**, 1183; *H. Monti, C. Corriol, M. Bertrand, Tetrahedron Lett.* **1982**, 947; *Ibid.* **1982**, 5539; *D. Solas, J. Wolinsky, J. Org. Chem.* **48**, 1988 (1983).*
- [7] *S. Takimoto, J. Inanaga, T. Katsuki, M. Yamagushi, Bull. Chem. Soc. Jpn.* **49**, 2335 (1976).
- [8] IR, ¹H-NMR and mass spectra are in perfect agreement with the assigned structure.
- [9] *R.W. Lang, H.-J. Hansen, Helv. Chim. Acta* **63**, 438 (1980). This preparation of the allenic ester **10** was superior to the approach described for the preparation of **3**: *M. Bertrand, Parfums, Cosmétiques, Aromes* **39**, 29 (1981).
- [10] *J.A. Dale, D.L. Dull, H.S. Mosher, J. Org. Chem.* **34**, 2543 (1969).
- [11] *W.H. Pirkle, J.R. Hauske, J. Org. Chem.* **42**, 1839 (1977).
- [12] The term "chiral efficiency" implies chemical yield and asymmetric induction of a reaction: *J.M. Wilson, D.J. Cram, J. Am. Chem. Soc.* **109**, 881 (1982).
- [13] The following compounds showed the indicated optical rotations [α]_D: **15**, -27.6° (c = 1.21, EtOH, 21.5°); synthetic (-)-**6**, -108.9° (c = 0.776, CHCl₃, 20°); natural (-)-**6**, -107.8° (c = 0.73, CHCl₃, 20°).
- [14] For highly asymmetric ene reactions and conjugate additions using chiral enoates see: *W. Oppolzer, C. Robbiani, K. Bättig, Helv. Chim. Acta* **63**, 2015 (1980); *Tetrahedron* **39** (1983) in press; *W. Oppolzer, H. Löher, Helv. Chim. Acta* **64**, 2808 (1981).

(Received in Germany 18 July 1983)